Machine Learning Algorithms for Predicting Complexation Properties of Weak Polyelectrolytes

Kartini Binti Ismail

Universiti Malaysia Perlis, Padang Besar Campus, Universiti Malaysia Perlis, Kampus Padang Besar, 02100 Padang Besar, Perlis, Malaysia.

Abdul Rahman Bin Ali

UCSI University, Terengganu Campus, UCSI University, Kampus Terengganu, Kuala Terengganu, Terengganu, Malaysia.

Abstract

Predicting the complexation properties of weak polyelectrolytes presents substantial challenges due to their partial ionization in solution and sensitivity to environmental conditions. The intricacies of these systems extend to the influence of molecular structure, size, and various other factors. To address this issue, we propose a comprehensive framework that leverages various machine learning algorithms, including regression models, decision trees, random forests, neural networks, support vector machines, Gaussian processes, k-nearest neighbors, and ensemble methods. The methodology involves several stages: data collection, feature engineering, model training, validation, testing, and interpretation. Data collected either from experiments or simulations are used to train the models, where features such as molecular weight, degree of ionization, and crosslink density are engineered to capture the essence of complexation behavior. The selected machine learning algorithms then facilitate the understanding and prediction of complexation properties under diverse conditions, including varying pH levels and ionic strengths. Importantly, we emphasize the critical role of domain-specific knowledge to interpret machine learning predictions effectively, ensuring they are aligned with physical and chemical principles. This multi-algorithmic approach offers an advanced toolset for the complexation study of weak polyelectrolytes, promising better predictive performance and interpretability than traditional methods.

Indexing terms: Weak Polyelectrolytes, Machine Learning Algorithms, Complexation Properties, Feature Engineering, Model Validation and Testing

Introduction

Predicting the complexation properties of weak polyelectrolytes presents an intricate task for researchers and engineers alike due to the multifaceted behavior of these macromolecules in various environmental settings [1]. Weak polyelectrolytes are polymers that can ionize partially when dissolved in a solution. Unlike strong polyelectrolytes, which are fully ionized under a wide range of conditions, the ionization degree of weak polyelectrolytes is sensitive to factors such as pH and ionic strength, thereby resulting in a dynamic charge state [2].

One of the most pivotal factors influencing the complexation properties of a system is the pH of the surrounding environment. Complexation refers to the formation of a complex between two or more species, often involving the interactions between a cation and an anion. The nature of this interaction can be profoundly impacted by the pH level because of its effect on the ionizable groups within the compounds, specifically within polymers. Polymers may contain ionizable groups such as carboxylic acids or amines that make them weak polyelectrolytes [3–5]. These weak polyelectrolytes are responsive to changes in pH, which can alter the state of these ionizable groups. For instance, an acidic environment might protonate the amine groups, whereas an alkaline environment could deprotonate the carboxylic acids. These changes in protonation or deprotonation significantly influence the ionization degree of the polyelectrolyte [6].

The ionization degree of a weak polyelectrolyte is pivotal for its electrostatic interactions with other species present in the solution. As the ionization degree changes, it affects the charge density of the polyelectrolyte, thereby modulating its interactions with other ions or molecules in the system. Electrostatic interactions are foundational to the complexation behavior, especially in biological and chemical systems where

specific complex formations are vital for numerous functionalities [7,8]. Therefore, when the pH alters the ionization state of the polyelectrolyte, it consequently impacts its ability to form complexes with other species. This variation in complexation behavior could be manifested in multiple ways, such as changes in solubility, stability, or reactivity of the formed complex [9].

Due to the sensitive nature of complexation to pH variations, making precise predictions regarding complexation becomes exceedingly complex. Even slight deviations in pH can yield considerably disparate behaviors. For example, a slight increase in pH might result in the decomplexation of previously stable complexes, or vice versa. This makes it challenging for scientists and researchers to formulate accurate models or frameworks for predicting complexation behavior in varying pH environments [10,11]. Hence, meticulous control and monitoring of pH levels are essential during experiments or industrial processes that involve complexation. This is especially pertinent in fields such as pharmaceuticals, water treatment, and biochemical engineering where slight changes can have far-reaching consequences [12].

Ionic strength is a key parameter that exerts a substantial influence on the complexation properties of solutions containing weak polyelectrolytes [13]. Ionic strength quantifies the concentration of ions in a solution and is defined as the sum of the concentrations of all the ionic species present, each multiplied by the square of their respective charges. In environments where the ionic strength is high, electrostatic forces between ions in the solution become shielded or screened, reducing the intensity of their interactions. As a result of this shielding effect, the ionization degree of weak polyelectrolytes tends to diminish. Less ionization directly translates to a lower density of charges on the polyelectrolyte chain, which consequently can lead to weaker electrostatic attractions with other ions or molecules in the solution [14]

The implications of reduced ionization due to increased ionic strength are particularly significant when considering the formation of complexes with other charged species. In solutions of high ionic strength, the diminished electrostatic forces can lead to less effective interactions between the weak polyelectrolyte and other charged species. Specifically, the charges on the polyelectrolytes and other ions can become screened, leading to a decrease in the strength and stability of the formed complexes. This is critical in various applications such as in pharmaceutical formulations, where the efficacy of a drug could be affected by its ability to form stable complexes, or in environmental science, where complexation can be crucial for pollutant removal.

Given the strong influence of ionic strength on complexation behavior, it becomes imperative for models designed to predict such behavior to accurately account for this parameter [15–17]. A predictive model that overlooks the role of ionic strength is likely to offer an unreliable estimation of complex stability or formation kinetics. Therefore, when constructing models to anticipate complexation properties, inclusion of terms that explicitly account for ionic strength is essential. This is particularly important for applications where precise control over complex formation is required, such as in bioengineering, water treatment processes, and chemical synthesis, where an inaccurate prediction could lead to undesirable outcomes or inefficiencies [18].

Moreover, the inherent molecular structure of the polyelectrolytes also plays a pivotal role in defining their complexation properties. Structural elements like molecular weight, branching, and presence of functional groups significantly influence the complexation behavior. Higher molecular weight often increases the availability of ionizable sites, whereas branching can either facilitate or impede complexation depending on the spatial arrangement of ionizable groups. Additionally, the presence of different functional groups can introduce more complexity as they might have distinct ionization constants, further adding to the variability in behavior.

Besides these factors, other variables like temperature, solvent quality, and concentration also contribute to the complexation properties. Due to the dependency on an array of factors, computational approaches for predicting the complexation behavior

often involve intricate mathematical models incorporating molecular dynamics or Monte Carlo simulations. Even then, the predictive power of these models can be limited and may require empirical adjustments [19–21].

Machine Learning Algorithms for Predicting

Machine learning (ML) has gained significant traction as an indispensable tool in the prediction and understanding of complex material behaviors. Traditionally, the investigation of material properties has relied on experimental methods, finite-element analysis, and computational modeling approaches such as density functional theory. While these techniques have their merits, they often fall short in tackling problems that involve multiple interacting variables, especially when the scale of data is enormous. Machine learning algorithms, specifically supervised learning techniques like support vector machines and neural networks, offer an effective alternative in these scenarios. They are capable of handling large datasets and can identify patterns or trends that are not readily apparent through traditional methods. ML models can learn from existing data to predict outcomes with high accuracy, which is crucial in areas such as materials science where small discrepancies can lead to significant material or process failures [22].

The applications of machine learning in materials science are diverse and far-reaching. One major area is the prediction of mechanical properties of alloys and composite materials. By learning from existing databases of material compositions, mechanical testing results, and microstructure images, ML algorithms can predict properties such as tensile strength, corrosion resistance, and thermal conductivity. This predictive capability is invaluable in speeding up the design and development process for new materials [23,24]. Additionally, machine learning can be integrated into real-time monitoring systems to predict material fatigue or failure, thereby serving as an early warning system to prevent catastrophic events in industrial applications [25].

However, it is essential to consider the limitations and challenges associated with employing machine learning in the materials science domain. The quality of predictions largely depends on the quality and size of the training data. Inaccurate or incomplete data can lead to erroneous predictions. Furthermore, interpretability remains a significant challenge; while ML algorithms can make accurate predictions, understanding the underlying physical mechanisms that led to those predictions is not straightforward. This makes it difficult to fully trust machine learning models in critical applications without robust verification and validation processes [26].

Regression models have been extensively employed in diverse scientific and engineering domains to establish relationships between input features and desired output variables. In the context of predicting complex material behaviors, linear regression provides an initial, yet simplistic, method for correlating input features like molecular structure, pH, and ionic strength with output variables such as the degree of complexation. However, the key limitation of linear regression is its inability to capture non-linear relationships that are often intrinsic to complex systems in materials science. On the other hand, polynomial regression offers the flexibility to model non-linear behaviors. While this is a strength, it is simultaneously a potential pitfall, as choosing an inappropriately high degree for the polynomial can result in overfitting, thereby reducing the model's predictive performance on new, unseen data [27].

Decision Trees and Random Forest models offer another layer of complexity and effectiveness in predicting complex material behaviors. These algorithms are particularly adept at capturing non-linear relationships between features and output variables. One of the principal advantages of decision trees and their ensemble version, the Random Forest, is their interpretability. Unlike many machine learning models where the decision-making process is often opaque, these algorithms allow for a clearer understanding of which features are most influential in making predictions. This attribute makes them highly valuable in fields like materials science, where understanding the relationship between material properties and underlying features can

be as important as making accurate predictions. These models are not only capable of high predictive accuracy but also provide insights into feature importance, aiding in the more focused design of experiments or subsequent data collection efforts.

Nevertheless, it is important to acknowledge the limitations and challenges that come with using regression models and tree-based methods in materials science. While decision trees and Random Forest models are interpretable, they can also become overly complex, leading to difficulties in model validation and generalization to new datasets [28,29]. Similarly, both linear and polynomial regression models are highly sensitive to the quality of the input data and may require preprocessing steps like normalization and feature selection to be effective [30].

Neural networks, particularly advanced variants such as feed-forward neural networks and convolutional neural networks (CNNs) [31], have become increasingly popular in the prediction and analysis of complex material behaviors. These algorithms excel at capturing complex non-linear relationships, making them ideally suited for problems in materials science that are too intricate for simpler models. The depth and structure of these networks allow them to automatically generate useful features from the data, which is especially beneficial when dealing with large datasets that contain a wealth of latent information. Deep learning models have shown impressive performance in applications ranging from the prediction of mechanical properties of alloys to real-time monitoring systems for material failure detection. However, these models come with challenges, most notably the need for large, high-quality datasets for training, and the inherent difficulty in interpreting the models. The so-called "black-box" nature of neural networks makes them less transparent in explaining why a particular prediction was made, which can be a significant drawback in critical applications where understanding the reasoning behind predictions is crucial [32].

Support Vector Machines (SVM) offer another machine learning approach to predicting complex material behaviors. One of the key strengths of SVM lies in its versatility, particularly through the use of different kernel functions, to model both linear and non-linear relationships in the data. In tasks such as classifying or regressing complexation behaviors based on input parameters like molecular structure or ionic strength, SVM can provide robust and accurate models. The kernel trick enables SVM to map input features to a higher-dimensional space, where a hyperplane can be optimally positioned to segregate or predict output variables. This makes SVM uniquely suited for problems that involve intricate boundaries between different classes or complex relationships between input and output variables.

Despite the strengths of neural networks and SVM, it's crucial to recognize their limitations for complete and effective application in materials science. For neural networks, the computational cost is often high, requiring specialized hardware like GPUs for training and inference. For SVM, the choice of an appropriate kernel function and the tuning of parameters like the regularization term are critical to the model's performance but can be computationally expensive and time-consuming. To address these challenges, ongoing research is increasingly focusing on model optimization, hybrid approaches that combine the benefits of multiple algorithms, and the development of interpretability techniques to make these powerful tools more transparent and reliable for predicting complex material behaviors [33].

Gaussian Processes (GPs) serve as an important statistical tool for tackling regression problems, especially when there is a need to account for uncertainty in predictions. In the realm of materials science, where complex physical properties like tensile strength, thermal conductivity, or degree of complexation are often subject to variability, the ability of GPs to provide uncertainty estimates becomes significantly valuable. These uncertainty estimates can help in making informed decisions, such as whether further experimentation is needed or if the predicted value is sufficiently reliable for practical application. Gaussian Processes are also non-parametric, which means they are flexible in capturing complex relationships between variables without having to specify a predetermined form of the function that links the input and output. This flexibility allows them to be tailored for various kinds of data complexities encountered in materials science [34].

K-Nearest Neighbors (KNN) offers another machine learning approach that is particularly versatile, capable of handling both classification and regression tasks. The fundamental premise of KNN is to make predictions based on the similarity between instances in the dataset. In materials science, where identifying relationships between different properties or states can be critical, KNN can be particularly useful if there is a well-understood similarity metric for the dataset at hand. For example, in predicting the mechanical properties of alloys based on compositional data, KNN could provide effective models if the similarity between different alloy compositions is well-defined and meaningful [35,36]. It's a straightforward algorithm that is easy to implement and interpret, making it a useful starting point or supplementary method in complex material behavior prediction tasks [37].

However, it is crucial to recognize the limitations of Gaussian Processes and K-Nearest Neighbors for their application to be effective. Gaussian Processes can become computationally expensive as the size of the dataset grows, limiting their scalability. Meanwhile, KNN's performance is highly sensitive to the choice of the similarity metric and the value of k (the number of neighbors considered), and it can suffer from the curse of dimensionality in high-dimensional spaces. To ameliorate these challenges, there is ongoing research aimed at improving the computational efficiency of GPs through techniques like sparse approximations, and enhancing the robustness of KNN through feature selection and distance metric learning. These advances aim to bolster the effectiveness of these models in predicting complex material behaviors reliably and efficiently.

Implementation

The implementation of machine learning models for studying complex behaviors of weak polyelectrolytes necessitates a structured approach. The initial step is data collection, where experimental or simulation data related to weak polyelectrolyte complexation should be systematically gathered under various conditions. The quality of the data collected is crucial, as it will directly affect the predictive capability of the subsequent machine learning model. The data should ideally cover a wide range of conditions, such as different pH levels, ionic strengths, and concentrations, to ensure that the machine learning model will be applicable across diverse scenarios. Data integrity checks and preprocessing may also be necessary to remove outliers or anomalies that could skew the model's performance.

Feature engineering follows as the next crucial step, wherein meaningful features from the raw data are extracted or derived. In the context of polyelectrolytes, the features could include variables like molecular weight, degree of ionization, crosslink density, and other properties relevant to the study of complexation behaviors. Proper feature engineering is vital to capture the intricate relationships that exist in the complex systems of polyelectrolytes. It is a challenging but essential phase as selecting the right features will significantly affect the model's predictive capabilities. The use of domain knowledge is particularly valuable in this step to identify features that are both scientifically meaningful and computationally effective.

Upon the completion of feature engineering, the next step involves model training. A suitable machine learning model is selected based on the problem definition and the nature of the data. For example, if the focus is on predicting continuous outcomes, regression models like Linear Regression or Support Vector Machines may be considered. Alternatively, for classification tasks, methods like Decision Trees or Neural Networks might be more appropriate. The chosen model is then trained using a subset of the prepared data. Hyperparameter tuning and cross-validation are often employed during this phase to optimize the model's performance.

The fourth step entails model validation and testing to ensure that the model is robust and generalizable. Validation is typically performed on a separate subset of the data that was not used during the training phase to avoid overfitting. Techniques such as k-fold cross-validation are commonly employed to ensure that the model performs well on different subsets of the data. After validation, the model is tested on completely unseen data to evaluate its performance in real-world scenarios [38,39]. This step provides the necessary assurance of the model's reliability and readiness for practical applications [40].

Finally, model interpretation is undertaken to understand the model's predictions and to identify the importance of various features in relation to complexation properties. This is especially important in scientific applications like polyelectrolyte studies, where the objective is not just prediction but also understanding the underlying physical or chemical phenomena. Depending on the type of model used, different techniques for interpretation may be applied. For instance, Decision Trees and Random Forests offer inherent feature importance metrics, while for Neural Networks, techniques like LIME (Local Interpretable Model-agnostic Explanations) may be used. Understanding the influential features and their impact on complexation can offer insights for further experimental design or for refining existing theories related to polyelectrolyte behavior.

Conclusion

Predicting the complexation properties of weak polyelectrolytes is indeed a challenging task. These materials can partly ionize in solution, making their charge states highly dependent on factors such as pH and ionic strength. Additionally, the interactions of these polymers with other species in the solution can be influenced by a multitude of factors, including molecular structure and size. Given the multifaceted nature of these interactions, it becomes imperative to use sophisticated methods to model and predict complexation behaviors [41,42]. Traditional methods based on deterministic equations may fall short in capturing the complexities of these systems due to the significant amount of variability in the environmental conditions and the molecular characteristics of weak polyelectrolytes [43],

Machine learning (ML) algorithms have shown promise in predicting complex material behaviors by leveraging their capability to handle large and intricate datasets. Regression models such as Linear and Polynomial Regression can provide a foundational understanding of the relationship between the input parameters, like molecular structure, pH, and ionic strength, and the output parameters, such as the degree of complexation. Linear Regression serves as a starting point for capturing linear relationships, but it has limitations in handling non-linear behaviors. Polynomial Regression, while able to capture non-linear relationships, may become problematic due to overfitting when the polynomial degree is too high.

Decision Trees and Random Forests are other viable options. These models are capable of capturing non-linear behaviors and are interpretable, which aids in the understanding of the most influential features affecting complexation. The Random Forest algorithm, an ensemble of decision trees, can offer improved performance and reduced overfitting. Neural networks, particularly deep learning models like feed-forward neural networks or convolutional neural networks (CNNs), can manage complex non-linear relationships, making them potentially valuable in cases where large datasets are available [44].

Support Vector Machines (SVM) and Gaussian Processes offer additional avenues for exploration. SVMs can be employed with various kernel functions to classify or regress complexation behaviors based on input parameters. On the other hand, Gaussian Processes can be applied to regression problems and offer the added advantage of providing uncertainty estimates. This is especially beneficial when attempting to predict complex physical properties where uncertainty quantification is necessary for making informed decisions.

Other algorithms worth mentioning are K-Nearest Neighbors (KNN) and Ensemble Methods like gradient boosting or AdaBoost [45].

KNN operates based on similarity metrics and can be particularly effective if a welldefined similarity metric for the dataset exists. Ensemble methods combine several weak learners to produce a more robust model, thus offering another layer of complexity and adaptability to the prediction of complexation properties of weak polyelectrolytes. Overall, the landscape of machine learning offers a wide array of algorithms and techniques that can be tailored to tackle the intricate problem of predicting the complexation properties of weak polyelectrolytes [46].

The implementation of machine learning algorithms for predicting the complexation properties of weak polyelectrolytes involves a series of carefully orchestrated steps, each of which contributes to the overall effectiveness and reliability of the resulting model. The first step, Data Collection, is critical and necessitates gathering experimental or computational data that captures various aspects of weak polyelectrolyte complexation under different conditions. This could include datasets from spectroscopy experiments, molecular dynamics simulations, or other methodologies designed to measure or infer the complexation behavior of weak polyelectrolytes.

The next phase is Feature Engineering, which involves the transformation or extraction of meaningful features from the raw data. Given the complexity of weak polyelectrolyte systems, the selected features could range from molecular weight and degree of ionization to crosslink density and even more advanced parameters like interaction energies. This step is crucial for the machine learning model to adequately capture the complexities of the system and make accurate predictions. It might also involve normalization or standardization procedures to ensure that features are commensurate and can be effectively used by the machine learning algorithm.

The third step, Model Training, requires selecting an appropriate machine learning algorithm based on the problem's requirements and the nature of the data. A subset of the available data is used to train the chosen model, optimizing its parameters to best fit the observed complexation behaviors. Algorithms such as Decision Trees, Neural Networks, or Support Vector Machines may be used, depending on the specific needs of the problem and the nuances of the data. During this phase, techniques like cross-validation may be employed to enhance the model's generalizability [47,48].

Following the training process, Model Validation and Testing are carried out to assess the reliability and robustness of the trained model. Validation is conducted on a separate subset of the data that was not used during training, which helps in identifying any overfitting issues. Subsequently, the model is tested on entirely unseen data to evaluate its predictive accuracy and reliability. Metrics such as root mean square error, precision, and recall might be used to quantify the model's performance. The final step, Model Interpretation, focuses on deriving insights from the machine learning model to understand the importance of the selected features in predicting complexation properties. Feature importance scores, partial dependence plots, or other interpretability tools may be used depending on the type of model employed. This step is essential not only for validation purposes but also for gaining scientific insights that can guide future experimental or computational research on weak polyelectrolytes.

References

- [1] Choi J, Rubner MF. Influence of the degree of ionization on weak polyelectrolyte multilayer assembly. Macromolecules 2005;38:116–24. https://doi.org/10.1021/MA048596O.
- [2] Rathee V, Sidky H, Sikora B, Whitmer J. Modeling Explicit-Ion Effects in Weak Polyelectrolyte Complexes. APS March Meeting Abstracts, vol. 2019, 2019, p. F55-002.
- [3] Chung AJ, Rubner MF. Methods of Loading and Releasing Low Molecular Weight Cationic Molecules in Weak Polyelectrolyte Multilayer Films. Langmuir 2002;18:1176–83. https://doi.org/10.1021/la010873m.
- [4] Kirwan LJ, Papastavrou G, Borkovec M, Behrens SH. Imaging the Coil-to-Globule Conformational Transition of a Weak Polyelectrolyte by Tuning the Polyelectrolyte Charge Density. Nano Lett 2004;4:149–52. https://doi.org/10.1021/nl0349121.
- [5] Stornes M, Linse P, Dias RS. Monte Carlo Simulations of Complexation between Weak Polyelectrolytes and a Charged Nanoparticle. Influence of Polyelectrolyte Chain Length and Concentration. Macromolecules 2017;50:5978–88. https://doi.org/10.1021/acs.macromol.7b00844.
- [6] Whitmer JK, Rathee VS, Sikora B. Weak polyelectrolytes in Confined Geometries. APS March Meeting Abstracts, vol. 2017, 2017, p. S6-007.
- [7] Zhumadilova GT, Gazizov AD, Bimendina LA, Kudaibergenov SE. Properties of polyelectrolyte complex membranes based on some weak polyelectrolytes. Polymer 2001;42:2985–9. https://doi.org/10.1016/S0032-3861(00)00442-0.
- [8] Xu L, Pristinski D, Zhuk A, Stoddart C, Ankner JF. Linear versus exponential growth of weak polyelectrolyte multilayers: Correlation with polyelectrolyte complexes 2012.
- [9] Rathee V, Qu S, Dilenschneider T, Phillip WA, Whitmer JK. Thermodynamics of Ionic Transport through Functionalized Membranes. APS March Meeting Abstracts, vol. 2016, 2016, p. R42-006.
- [10] Blaakmeer J, Bohmer MR, Stuart MAC. Adsorption of weak polyelectrolytes on highly charged surfaces. Poly (acrylic acid) on polystyrene latex with strong cationic groups. Macromolecules 1990.
- [11] Kovačević D, van der Burgh S, de Keizer A, Cohen Stuart MA. Specific ionic effects on weak polyelectrolyte multilayer formation. J Phys Chem B 2003;107:7998–8002. https://doi.org/10.1021/jp0273777.
- [12] Rathee V. Mathematical Models for Evaporation of Thin Films 2013.
- [13] Liang W, Fan Z, Liang Y, Jia J. Cross-Attribute Matrix Factorization Model with Shared User Embedding. ArXiv Preprint ArXiv:230807284 2023.
- [14] Rathee V, Zervoudakis A, Sidky H, Sikora B, Whitmer J. Charging Behavior of Associating Weak Polyelectrolytes. APS March Meeting Abstracts, vol. 2018, 2018, p. H52-012.
- [15] Tanchak OM, Barrett CJ. Swelling dynamics of multilayer films of weak polyelectrolytes. Chem Mater 2004.
- [16] Ulrich S, Laguecir A, Stoll S. Complexation of a Weak Polyelectrolyte with a Charged Nanoparticle. Solution Properties and Polyelectrolyte Stiffness Influences. Macromolecules 2005;38:8939–49. https://doi.org/10.1021/ma051142m.
- [17] Xie AF, Granick S. Local Electrostatics within a Polyelectrolyte Multilayer with Embedded Weak Polyelectrolyte. Macromolecules 2002;35:1805–13. https://doi.org/10.1021/ma011293z.
- [18] Cortés-Morales EC, Rathee VS, Ghobadi A, Whitmer JK. A molecular view of plasticization of polyvinyl alcohol. The Journal of Chemical Physics 2021;155.
- [19] Nap RJ, Park SH, Szleifer I. Competitive calcium ion binding to end-tethered weak polyelectrolytes. Soft Matter 2018;14:2365–78. https://doi.org/10.1039/c7sm02434g.
- [20] Yuan W, Weng G-M, Lipton J, Li CM, Van Tassel PR, Taylor AD. Weak polyelectrolyte-based multilayers via layer-by-layer assembly: Approaches, properties, and applications. Adv Colloid Interface Sci 2020;282:102200. https://doi.org/10.1016/j.cis.2020.102200.

32|PAGE INTERNATIONAL JOURNAL OF APPLIED MACHINE LEARNING AND COMPUTATIONAL INTELLIGENCE

- [21] Plamper FA, Ruppel M, Schmalz A, Borisov O. Tuning the Thermoresponsive Properties of Weak Polyelectrolytes: Aqueous Solutions of Star-Shaped and Linear Poly(N,N-dimethylaminoethyl Methacrylate) 2007.
- [22] Liang Y, Liang W. ResWCAE: Biometric Pattern Image Denoising Using Residual Wavelet-Conditioned Autoencoder. ArXiv Preprint ArXiv:230712255 2023.
- [23] Uhlík F, Košovan P, Limpouchová Z, Procházka K, Borisov OV, Leermakers FAM. Modeling of Ionization and Conformations of Starlike Weak Polyelectrolytes. Macromolecules 2014;47:4004-16. https://doi.org/10.1021/ma500377y.
- [24] Plamper FA, Ballauff M, Müller AHE. Tuning the Thermoresponsiveness of Weak Polyelectrolytes by pH and Light: Lower and Upper Critical-Solution Temperature of Poly(N,N-dimethylaminoethyl Journal of the American 2007.
- [25] Rathee VS. Modeling Weak Polyelectrolyte Behavior and Predicting Complexation Properties. University of Notre Dame; 2019.
- [26] Liang Y, Liang W, Jia J. Structural Vibration Signal Denoising Using Stacking Ensemble of Hybrid CNN-RNN. ArXiv E-Prints 2023:arXiv-2303.
- [27] Liang W, Liang Y, Jia J. MiAMix: Enhancing Image Classification through a Multi-stage Augmented Mixied Sample Data Augmentation Method. ArXiv Preprint ArXiv:230802804 2023.
- [28] Elzbieciak M, Kolasińska M, Zapotoczny S, Krastev R, Nowakowska M, Warszyński P. Nonlinear growth of multilayer films formed from weak polyelectrolytes. Colloids Surf A Physicochem Eng Asp 2009;343:89-95. https://doi.org/10.1016/j.colsurfa.2009.01.034.
- [29] Landsgesell J, Holm C, Smiatek J. Simulation of weak polyelectrolytes: a comparison between the constant pH and the reaction ensemble method. Eur Phys J Spec Top 2017;226:725-36. https://doi.org/10.1140/epjst/e2016-60324-3.
- [30] Rathee VS, Sidky H, Sikora BJ, Whitmer JK. Role of associative charging in the entropy-energy balance of polyelectrolyte complexes. J Am Chem Soc 2018;140:15319-28.
- [31] Liang Y. Structural Vibration Signal Denoising Using Stacking Ensemble of Hybrid CNN-RNN. Advances in Artificial Intelligence and Machine Learning. 2022; 3 (2): 65 2006.
- [32] Chavez A, Koutentakis D, Liang Y, Tripathy S, Yun J. Identify statistical similarities and differences between the deadliest cancer types through gene expression. ArXiv Preprint ArXiv:190307847 2019.
- [33] Zhu Y, Yan Y, Zhang Y, Zhou Y, Zhao Q, Liu T, et al. Application of Physics-Informed Neural Network (PINN) in the Experimental Study of Vortex-Induced Vibration with Tunable Stiffness. The 33rd International Ocean and Polar Engineering Conference, OnePetro; 2023.
- [34] Wu X, Bai Z, Jia J, Liang Y. A Multi-Variate Triple-Regression Forecasting Algorithm for Long-Term Customized Allergy Season Prediction. ArXiv Preprint ArXiv:200504557 2020.
- [35] Panagiotopoulos AZ. Charge correlation effects on ionization of weak polyelectrolytes. Phys Condens Matter 2009;21:424113. J https://doi.org/10.1088/0953-8984/21/42/424113.
- [36] Eriksson M, Notley SM, Wågberg L. The influence on paper strength properties when building multilayers of weak polyelectrolytes onto wood fibres. J Colloid Interface Sci 2005;292:38-45. https://doi.org/10.1016/j.jcis.2005.05.058.
- [37] Rathee VS, Zervoudakis AJ, Sidky H, Sikora BJ, Whitmer JK. Weak polyelectrolyte complexation driven by associative charging. J Chem Phys 2018;148.
- [38] Mauser T, Déjugnat C, Möhwald H, Sukhorukov GB. Microcapsules made of weak polyelectrolytes: templating and stimuli-responsive properties. Langmuir 2006;22:5888-93. https://doi.org/10.1021/la060088f.
- [39] Kharlampieva E, Sukhishvili SA. Ionization and pH Stability of Multilayers Formed by Self-Assembly of Weak Polyelectrolytes. Langmuir 2003;19:1235–43. https://doi.org/10.1021/la026546b.
- [40] Rathee VS, Sidky H, Sikora BJ, Whitmer JK. Explicit ion effects on the charge and conformation of weak polyelectrolytes. Polymers 2019;11:183.

- [41] Nová L, Uhlík F, Košovan P. Local pH and effective p KA of weak polyelectrolytes-insights from computer simulations. Phys Chem Chem Phys 2017.
- [42] Evers OA, Fleer GJ, Scheutjens JMHM, Lyklema J. Adsorption of weak polyelectrolytes from aqueous solution. J Colloid Interface Sci 1986;111:446–54. https://doi.org/10.1016/0021-9797(86)90047-0.
- [43] Cortes Morales EC, Whitmer J, Rathee V, Ghobadi A. Molecular view of the plasticization process of poly (vinyl) alcohol. APS March Meeting Abstracts, vol. 2022, 2022, p. G00-107.
- [44] Hernandez L, Hassan M, Shukla VP. Applications of Cloud Computing in Intelligent Vehicles: A Survey. Journal of Artificial Intelligence and Machine Learning in Management 2023;7:10–24.
- [45] Bai Z, Yang R, Liang Y. Mental task classification using electroencephalogram signal. ArXiv Preprint ArXiv:191003023 2019.
- [46] Rathee VS, Qu S, Phillip WA, Whitmer JK. A coarse-grained thermodynamic model for the predictive engineering of valence-selective membranes. Molecular Systems Design & Engineering 2016;1:301–12.
- [47] Nap R, Gong P, Szleifer I. Weak polyelectrolytes tethered to surfaces: Effect of geometry, acid–base equilibrium and electrical permittivity. J Polym Sci B Polym Phys 2006;44:2638–62. https://doi.org/10.1002/polb.20896.
- [48] Kovacevic D, van der Burgh S, de Keizer A, Cohen Stuart MA. Kinetics of Formation and Dissolution of Weak Polyelectrolyte Multilayers: Role of Salt and Free Polyions. Langmuir 2002;18:5607–12. https://doi.org/10.1021/la025639q.